<u>Description</u>: This course is designed to create foundational knowledge and proficiency in essential chemistry laboratory techniques and skills. It includes hands-on experiments designed to illustrate the relationships between the structures of compounds and their resulting properties. It is also intended to provide students with additional exposure/experience with lab safety and proper handling of chemicals. Lastly, the lab will introduce and develop techniques for purifying chemical compounds and determining their structures.

## Course Objectives:

- 1. To establish best practices for working safely in lab—including maintaining good chemical hygiene, wearing personal protective equipment, knowing the locations of and proper use of safety showers, eye washes, fire extinguishers, etc., maintaining good ventilation, and proper waste disposal.
- 2. To develop information literacy skills by searching for authoritative, reliable physical property data from sources such as SciFinder Scholar, the Merck Index, PubChem, etc.
- 3. To develop analytical reasoning skills by determining the identities of unknown compounds using a combination of physical property determinations and various forms of spectroscopy.
- 4. To utilize experiments to demonstrate relationships between the structures of compounds and their properties, to develop laboratory techniques for measuring mass, volume, and temperature, and to gain practice in collecting, analyzing, and presenting data, including dealing with sources of error.
- 5. To separate mixtures using differences in chemical and physical properties and to analyze the composition of a mixture, to develop laboratory skills such as dissolving solids and preparing solutions of known concentration, to explore relationships between the structures of compounds and their partitioning behavior.

<u>Academic Calendar</u>: It is the student's responsibility to know the course schedule, posted on Sakai, and the official University Academic Calendar.

Meeting Times and Locations: All sections of CHEM 181 meet in LSB 115.

| Section | Day and Time                 | Instructor | Teaching Assistants                      |
|---------|------------------------------|------------|--|
| 181-001 | Mondays 8:30 AM - 11:15 AM   | Dr. Balija | Eric Reyes, Diman Hosseini               |
| 181-002 | Tuesdays 2:30 PM – 5:15 PM   | Dr. Balija | Emily Dominique, Diman Hosseini          |
| 181-003 | Tuesdays 5:30 PM – 8:15 PM   | Mr. Thomas | Claire Kennedy, Hussein Qtaish           |
| 181-004 | Thursdays 8:30 AM – 11:15 AM | Dr. Balija | Emily Radz, Claire Kennedy               |
| 181-005 | Thursdays 11:30 AM - 2:15 PM | Dr. Basner | Emily Radz, Claire Kennedy               |
| 181-006 | Thursdays 2:30 PM – 5:15 PM  | Dr. Basner | Emily Dominique, Eric Reyes              |
| 181-007 | Thursdays 5:30 PM – 8:15 PM  | Dr. Basner | Emily Dominique, Elizabeth Serna-Sanchez |
| 181-008 | Fridays 8:30 AM – 11:15 AM   | Dr. Balija | Eric Reyes, Elizabeth Serna-Sanchez      |
| 181-009 | Fridays 2:45 PM – 5:30 PM    | Dr. Basner | Ellie Sharp, Emily Radz                  |
| 181-010 | Fridays 5:45 PM – 8:30 PM    | Dr. Basner | Erwin Weerawardhana, Hussein Qtaish      |
| 181-011 | Saturdays 9:30 AM – 12:15 PM | Dr. Basner | Elizabeth Serna-Sanchez, Ellie Sharp     |

Office Hours: Office hours for the instructor and TAs will be posted on Sakai.

## Required Items:

- 1. Bound composition notebook (not spiral bound and with no tear-out perforations).
- 2. Access to Microsoft 365. This is provided by Loyola to students.
- 3. Safety goggles (We provide a pair to you on the first day). If you wish to get your own, they must be Type G, H, or K goggles and must meet the requirements of ANSI Z87.1.
- 4. Long-sleeved, full-length laboratory coat.
- 5. Appropriate clothing and footwear as described in the laboratory safety rules.
- 6. A non-erasable, waterproof pen.
- 7. Access to Sakai.

This course contains elements that require a desktop or laptop computer with high-speed Internet access. Some of the virtual lab simulations used MAY NOT WORK on tablets or mobile devices. If you do not have a desktop/laptop computer or Internet service, you will need to go to the Information Commons on campus or

contact the extended loan equipment program within the first few days of the start of the course and arrange for these resources.

<u>Course Homepage</u>: Announcements, assessments, extra copies of the handouts, the grade book, etc. are posted on <u>Sakai.luc.edu</u>. Students should check Sakai frequently as it is central to how the course operates. **Please note that all course materials should be accessed using the Lessons tab**, where details are broken down by topic/experiment. Some assignments may not open properly if accessed through other tabs.

Safety Rules: Students must read and agree to follow the lab safety rules before they can work in the lab. On lab safety day, students must sign a Lab Safety Contract that acknowledges that the student received the safety rules and that the student agrees to follow them. A lab safety contract must be signed by a student before they can work in the lab. Anyone who does not adhere to the safety rules will receive point deductions and may not be allowed to remain in the laboratory, depending on the severity of the violation. Students must bring eye protection and a full-length lab coat to every experiment. Students must also dress in appropriate clothing and footwear such that there is no exposed skin at any point below the shoulders. For the sake of hygiene and other reasons, students may not borrow goggles and/or a lab coat. Any student who comes to lab without these items will automatically not be allowed to perform the experiment. More information on the course attendance policy and safety points can be found below.

## \*\*\* WHENEVER CHEMICALS ARE PRESENT, NO ONE MAY ENTER LSB-115 UNLESS THEY ARE WEARING THE FOLLOWING: \*\*\*

- 1. EYE PROTECTION (These must be Type G, H or K goggles and must meet or exceed ANSI Z87.1)
- 2. FULL-LENGTH LAB COAT
- 3. CLOSED-TOE, CLOSED-HEEL SHOES
- 4. APPROPRIATE ATTIRE THAT FULLY COVERS ALL SKIN BELOW THE WAIST

<u>Grading</u>: Course grades consist of the following components:

| Pre-lab Exercises, drop lowest one        | 20%        |
|---|------------|
| Notebook Entries/Results, drop lowest one | 20%        |
| Unknowns, drop lowest one                 | 10%        |
| Post-lab Exercises, drop lowest one       | 20%        |
| Lab Reports                               | 10%        |
| Safety Points                             | 10%        |
| Final Lab Notebook Evaluation             | <u>10%</u> |
|   | 100% total |

A>93%, A->90%, B+>87%, B>83%, B->80%, C+>77%, C>73%, C->70, D+>67%, D≥60%, F<60%

Attendance: Students are expected to attend every lab session. Any student who does not have their safety goggles and lab coat, is not dressed appropriately, or has not completed the pre-lab preparation on time will be marked absent. Missing a lab period for any reason will result in an automatic zero for any work that is not completed. Students will be allowed to make alternate arrangements, which vary by experiment, for the points from ONE absence from lab during the course. Absent students are responsible for requesting the alternate arrangements via an email to the instructor within 48 hours of the absence. Any additional missed work beyond one experiment cannot be made up and will result in scores of zero.

There will be an attendance sheet students must sign upon entering the lab. The attendance sheet must match who is present in the lab in an emergency. If anyone must leave the lab after signing in (e.g., to use the restroom, get a drink of water, etc.) that person must be sure to talk to their TA to log out on the attendance sheet. For safety's sake, to obtain better results and to be fair to both lab partners, students should limit time out of the lab. Students who leave the lab for a period longer than 10 minutes will receive a deduction from the safety points for that experiment.

Additionally, to ensure fairness to everyone, students who arrive late may be asked to perform an experiment solo. Students must arrive within 15 minutes of the start time for lab to do the experiment; any student arriving more than 15 minutes late will be marked absent.

<u>Pre-Lab Preparation</u>: Success in lab depends on advance preparation. Students who come to class prepared get better results, get done faster, and, most importantly, tend to have fewer accidents. Therefore, there are several things that students must do BEFORE coming to the lab. One major component of the pre-lab assignment is to thoroughly examine the materials posted about the experiment on Sakai. Before coming to class, students must complete all the materials included in the "Pre-lab Preparation" section of Sakai for that experiment. All handouts must be opened and read. Videos must be viewed to 100% completion. Any Labster simulations must be completed to 100% progress. Pre-lab Exercises, which have unlimited attempts until closed, must be submitted at least once before the due date. To allow time for the pre-labs to be checked, the due date for the pre-lab preparation will be 30 minutes before class begins.

No pre-labs will be accepted less than 30 minutes before class, and students who do not complete the pre-lab preparation will not be allowed to perform the experiment.

Notebook/Lab Results: The ability to keep good records is a valuable skill. Proper documentation of experiments will make experimental results easier to interpret and, very importantly, will facilitate replication of the experiment. (Reproducibility is central to the scientific method!) One of the most important facets of experimental work is that everything should be recorded as completely and accurately as possible. Sometimes, important discoveries are made when things do not go exactly as expected. Therefore, it is critical that students report their actual data and not what it is thought that the correct answer should be. THE NOTEBOOK MUST ACCURATELY REFLECT WHAT HAPPENED DURING THE EXPERIMENT. Procedures should be written entirely in past tense to document the experiment as it was performed. Lab notebook entries are not meant to be written as instructions for others, but as a written record of what happened while performing the experiment.

For this course, notebook entries will be completed DURING LAB in a bound composition book or a similar bound notebook. Notebook entries will include the title, objective, a **detailed procedure section written while the experiment is being performed,** and a summary of the experimental results (usually as a table). The material normally covered in the discussion or conclusion section of the notebook will be included in the post-labs. Therefore, the discussion/conclusion section does not have to be included in the lab notebook for this class. After an is completed, the notebook pages will be uploaded on Gradescope before leaving lab. To receive credit, the pages must be legible and oriented properly.

<u>Unknowns</u>: Several of the experiments have an unknown that students must identify. Unknowns are randomly assigned and differ from student to student. Points are awarded on an all-or-nothing basis; there is no partial credit.

<u>Post-lab Exercises</u>: While performing an experiment, students are often focused on the tasks being completed and do not always pause to consider why certain things are done in specific ways. To reinforce the concepts and techniques that were performed in lab and to allow more time to think through the experiments, students will also complete post-lab exercises outside of class. Post-lab questions will be posted on Sakai. Students are allowed unlimited attempts until the due date, and assessments must be submitted to count. Spelling, grammar, and significant figures count.

<u>Lab Reports</u>: Students will compose two type-written lab reports that convey the results and importance of the experiments, "Synthesis of Benzoic Acid" and ". A lab report formalizes one's experiment with written documentation that is accurate and understandable to others.

<u>Lab Safety Points</u>: Laboratory safety is an extremely serious and important topic. All violations of the safety rules will result in point deductions. Some safety violations may also result in the student being expelled from the lab. As employees, the instructors and Teaching Assistants are expected to enforce the safety rules and disciplinary actions may be taken against them by Loyola if the safety rules are not enforced. Therefore, please do not ask the instructors or Teaching Assistants to ignore any safety rules or to not apply any penalties for safety infractions. They are not optional. Some examples of safety violations that result in immediate point deductions include things like removing safety goggles in the lab, touching your phone with your gloved hands,

eating, or drinking in lab, etc. These are just some examples—the list is not all-inclusive because it is impossible to foresee every potential safety violation. Some examples of safety violations that will result in immediate expulsion from the laboratory include things like wearing inappropriate attire, dumping chemical waste down the sink, etc. Again, these are just some examples. There are other safety violations that could come up that may result in a student being asked to leave the lab. In addition, incurring three, lesser safety violations in one lab period will result in a student being asked to leave the lab, even if the infractions would not warrant expulsion individually. Any student removed from lab for safety violations twice will automatically receive one lower letter grade. If a student is removed from lab for safety violations a third time, they will automatically fail the course. Each experiment will have three safety points at stake.

<u>Final Lab Notebook Evaluation</u>: Notebooks will be collected at the end of the semester and compared against the scanned notebook pages submitted on Sakai. Points up to and including full credit may be deducted if the actual notebook does not match the submitted files, does not match the characteristics described in the handout posted on Sakai, such as permanently bound, written in pen, etc. Notebook point deductions may also be made for not writing in third person or passive voice, using the wrong tense, grammatical errors, spelling errors, etc.

Re-grades: All requests to have any submitted assignment re-graded must be submitted in writing before the hard deadline. Students should email the instructor, not the TA.

<u>Late Policy</u>: Unless otherwise specified above, assignments may be submitted late. Assignments submitted late but within a week of the due date will automatically receive a 25% deduction. Assignments submitted beyond one week late will automatically receive a 50% deduction. Work submitted after the hard deadline will not be graded.

<u>Hard Deadline</u>: All materials of any kind must be submitted by 5 PM on Friday, April 26, 2024. No materials will be accepted after this time. Final grades will be calculated based only on materials submitted by this deadline. If there are substantial materials missing that cannot be submitted before this deadline, the student should consider withdrawing from the course or requesting an Incomplete by completing this form before the end of the term.

Email: Faculty email addresses are posted on the open Internet for every software bot and spammer in the world to see. Therefore, faculty Outlook accounts are configured differently, and an outside contractor also scans faculty email. Emails from outside sources are often blocked automatically. Because of this and a federal law relating to student privacy (FERPA), students must use a Loyola email address when contacting the TAs or the instructor about this course. In the subject line of an email, please put Chem 181-section number and TA's name.

Interactions with TAs: To increase the amount of individual assistance you receive in lab, Teaching Assistants will participate in delivering this course. If at any time during the semester you have any questions or concerns about the behavior of your Teaching Assistant, please contact the instructor.

Academic Integrity: All students in this course are expected to have read and to abide by the demanding standard of personal honesty, drafted by the College of Arts & Sciences, which can be viewed at:

http://www.luc.edu/cas/advising/academicintegritystatement/

"A basic mission of a university is to search for and to communicate the truth as it is honestly perceived. A genuine learning community cannot exist unless this demanding standard is a fundamental tenet of the intellectual life of the community. Students of Loyola University Chicago are expected to know, to respect, and to practice this standard of personal honesty. Academic dishonesty can take several forms, including, but not limited to cheating, plagiarism, copying another student's work, and submitting false documents."

Regarding the use of Artificial Intelligence: our Provost has expressed to "Let us all make sure we are learning and sharing best practices and not allowing AI to do the learning for us." In this course, any work you submit

for credit must represent your own ideas and understanding of the assigned material. If you are uncertain about any case where your use of AI may conflict with University or course standards, please see your instructor to discuss your concerns.

Any instance of dishonesty (including those detailed on the website provided above) will be reported to The Chair of The Department of Chemistry & Biochemistry, who will decide what the next steps may be. The penalty may include a grade of zero for that assignment and/or failure of the course.

<u>Health, Safety, and Well-Being On-Campus:</u> Please be familiar with and adhere to all policies and protocols posted on the Campus Info & Resources site: https://www.luc.edu/healthsafetyandwellbeing/campusinforesources/

Course/Instructor Evaluation – SmartEval: The following information came from the University regarding course evaluations, "Towards the end of the course, the students will receive an email from the Office of Institutional Effectiveness reminding them to provide feedback on the course. They will receive consistent reminders throughout the period when the evaluation is open, and the reminders will stop once they have completed the evaluation.

- -The evaluation is completely anonymous. When the results are released, instructors and departments will not be able to tell which student provided the individual feedback.
- -Because it is anonymous and the results are not released to faculty or departments until after grades have been submitted, the feedback will not impact a student's grade.
- -The feedback is important so that the instructor can gain insight into how to improve their teaching and the department can learn how best to shape the curriculum."

<u>Course Repeat Rule</u>: Effective with the Fall 2017 semester, students are allowed only THREE attempts to pass Chemistry courses with a C- or better grade. The three attempts include withdrawals (W). The Department advises that it is preferable to complete a course with a grade of C or C-, and to demonstrate growth in future coursework, then to withdraw from a course.

After the second attempt, the student must secure approval for a third attempt. Students must come to the Chemistry Department, fill out a permission to register form or print it from the Department of Chemistry & Biochemistry website: <a href="https://www.luc.edu/chemistry/forms/">https://www.luc.edu/chemistry/forms/</a> and personally meet and obtain a signature from either the Undergraduate Program Director, Assistant Chairperson, or Chairperson in Chemistry. A copy of this form is then taken to your Academic Advisor in Sullivan to secure final permission for the attempt.

Student Accommodations: Loyola University Chicago provides reasonable accommodations for students with disabilities. Any student requesting accommodations related to a disability or other condition is required to register with the Student Accessibility Center (SAC). Professors will receive an accommodation notification from SAC, preferably within the first two weeks of class. Students are encouraged to meet with their professor individually to discuss their accommodations. All information will remain confidential. Please Note that in this class, software may be used to audio record class lectures to provide equitable access to students with disabilities. Students approved for this accommodation use recordings for their personal study only and recordings may not be shared with other people or used in any way against the faculty member, other lecturers, or students whose classroom comments are recorded as part of the class activity. Recordings are deleted at the end of the semester. For more information about registering with SAC or questions about accommodations, please contact SAC at 773-508-3700 or SAC@luc.edu.

<u>Loyola University Absence Policy for Students in Co-Curricular Activities (including ROTC)</u>: Students missing classes while representing Loyola University Chicago in an official capacity (e.g., intercollegiate athletics, debate team, model government organization) shall be allowed by the faculty member of record to make up any assignments and to receive notes or other written information distributed in the missed classes.

Students should discuss with faculty the potential consequences of missing lectures and how they can be remedied. Students must provide their instructors with proper documentation, i.e., <u>"Athletic Competition & Travel Letter"</u> describing the reason for and date of the absence.

Chem 181 Reactivity Lab

This documentation must be signed by an appropriate faculty or staff member, and it must be provided to the professor in the first week of a semester. It is the student's responsibility to make up any assignments. If the student misses an examination, the instructor must allow them to take it at another time. (https://www.luc.edu/athleteadvising/attendance.shtml)

Students who will miss class for an academic competition or conference must provide proper documentation to their instructor as early in the semester as possible.

Accommodations for Religious Reasons: If you have observances of religious holidays that will cause you to miss class or otherwise effect your performance in the class you must alert the instructor within 10 calendar days of the first class meeting of the semester to request special accommodations, which will be handled on a case-by-case basis.

<u>Privacy Statement</u>: Assuring privacy among faculty and students engaged in online and face-to-face instructional activities helps promote open and robust conversations and mitigates concerns that comments made within the context of the class will be shared beyond the classroom. As such, recordings of instructional activities occurring in online or face-to-face classes may be used solely for internal class purposes by the faculty member and students registered for the course, and only during the period in which the course is offered. Students will be informed of such recordings by a statement in the syllabus for the course in which they will be recorded. Instructors who wish to make subsequent use of recordings that include student activity may do so only with informed written consent of the students involved or if all student activity is removed from the recording. Recordings including student activity initiated by the instructor may be retained by them only for individual use. Additionally, all materials from this course cannot be shared outside the course without the instructor's written permission.

Instructor Contacts: Mr. Thomas, LSB 124, (773) 508-8115, tthoma1@luc.edu

Dr. Basner, FH 428, <u>abasner@luc.edu</u> Dr. Balija, FH 104, <u>abalija@luc.edu</u>

## **Experiments**

- 1. Chemistry Information & Software Resources
- 2. Laboratory Safety
- 3. Molecular Structure and Relative Acid Strength
- 4. Determination of the pKa and Molar Mass of Nicotinic Acid by Titration
- 5. Boiling Point Determination and Infrared Spectroscopy
- 6. Mass Spectrometry and C13 NMR Spectroscopy
- 7. Distillation
- 8. Extraction
- 9. Thin-Layer Chromatography (TLC)
- 10. Kinetics of Nucleophilic Substitution
- 11. Synthesis of Benzoic Acid
- 12. Natural Product Isolation
- 13. Ketone Derivatives